

# Sustainable Management of Red Mud: Lifecycle Assessment and Treatment Techniques

Remisha S.R<sup>1</sup>, Reshmi R<sup>2</sup>, Nithyalakshmi B<sup>3</sup>

<sup>1,2</sup>PG Scholar, Department of Civil Engineering Kumaraguru College of Technology, Coimbatore.

<sup>3</sup>Assistant Professor, Department of Civil Engineering Kumaraguru College of Technology, Coimbatore.

## ABSTRACT

Red mud is a strongly alkaline leaching solid hazardous waste material generated in the Bayer process of alumina production. India ranks 4<sup>th</sup> place in alumina production. According to the Ministry of Mines (Oct 2022), India produces 9 million tonnes of red mud annually. In nature, red mud is highly alkaline. The high alkalinity and caustic content of red mud will contaminate the groundwater and fertile soil, and also there is a limitation on the utilization of red mud. Due to its low utilization ratio, large amounts of red mud are stacked in disposal areas, which may cause water pollution, soil salinization and heavy metal pollution. Red mud contains toxic metals such as arsenic, chromium, lead, nickel, zinc and radioactive elements such as thorium, potassium, and uranium, which cause severe effects on the environment and also on our health. The objective is to examine the environmental impact, utilization methods and treatment process of the red mud (bauxite residue). Also, scrutinize the biodegradation of heavy metals in red mud.

**Keywords:** Red mud, Bayer process, toxic metals, radioactive materials.

## I INTRODUCTION

Aluminum plays a vital role in day-to-day life. Aluminium is widely used in cans, foils, kitchen utensils, aeroplane parts. After oxygen, silica aluminium is third most abundant material in lithosphere. In 1854, the first production of aluminium is started. There are two steps in the alumina production one is Bayer process and other is hall-Heroult process. In the Bayer process, filtration after the addition of sodium hydroxide produces red mud. 1 ton of alumina production gives 1.5 tons of red mud. Currently about 4.6 billion tons of red mud is stored with an increase of 200mt per annum. Due to its high alkalinity, it causes serious effects to the environment such as human health effects due to the presence of heavy metals, soil pollution, leachate will also affect the ground water.

There is no proper utilization or treatment technique in India. The red mud is disposed as slurry form in sea or lake or it is disposed in dry form in land. In this paper the properties, environmental effects and its disposal and treatment techniques is highlighted.

## II BAUXITE:

Bauxite is a rock formed in the laterite soil. The French geologist Pierre Berthier named bauxite in 1821. bauxite contains clay minerals, aluminium hydroxide, hydrous aluminium oxides. The insoluble minerals present in bauxite is hematite, quartz, magnetite, siderite, goethite. The specific gravity of bauxite is 2.6 to 3.5.



**Fig 1 Bauxite**

Bauxite ore differ three types according to its crystallography. They are Gibbsitic ( $\gamma\text{-Al}(\text{OH})_3$ ), Boehmitic ( $\gamma\text{AlO}(\text{OH})$ ), Diasporic ( $\alpha\text{-AlO}(\text{OH})$ ).

BAXUITE ORE TYPE	COUNTRY
Gibbsitic	Australia, Brazil, Ghana, Guyana, India (eastern coast), Indonesia, Jamaica, Malaysia, Sierra leone, Suriname, Venezuela
Boehmitic	Australia, Guinea, Hungary, USSR, Yugoslavia, India (Central part)
Diasporic	China, Greece, Guinea, Romania, Turkey

Table 1 Bauxite ore of different country

### III BAUXITE DEPOSITS OF INDIA

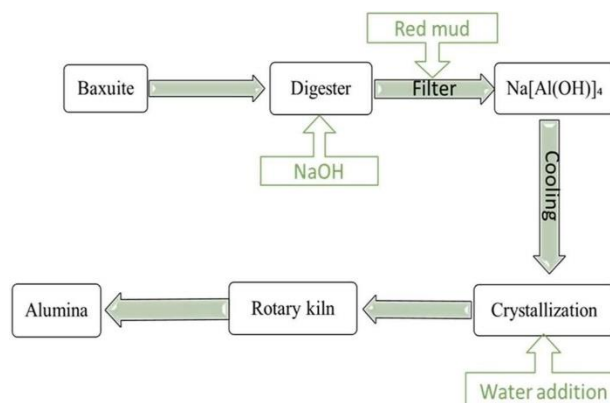
Odisha produces country’s 52% of bauxite resource, Andhra Pradesh produces 18%, Gujarat 7%, Chhattisgarh and Maharashtra 5%, Madhya Pradesh and Jharkhand 4%.



**Fig 2 Aluminium plants in India**

### IV RED MUD:

#### A. Production process:



**Fig 3 Bayer process**

**V CHARACTERISTICS OF RED MUD:**

**A. Chemical characteristics:**

COMPANY	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	TiO <sub>2</sub>	SiO <sub>2</sub>	CaO	Na <sub>2</sub> O	P <sub>2</sub> O <sub>5</sub>	V <sub>2</sub> O <sub>3</sub>	LOI
BALCO	21.9	28.1	15.6	7.5	10.2	4.5	-	-	12.2
HINDALCO	24.3	24.5	18	6.2	-	5.3	-	-	-
NALCO	14.8	54.8	3.7	6.4	2.5	4.8	0.67	0.38	9.5
INDAL	19.4	27.9	16.4	7.3	16.4	11.8	3.3	-	12.6
MALCO	14	18	50	56	2-4	6-9	1-2	-	12.6

Table 2 Chemical concentrations from different companies in India

**B. Physical and geo technical properties:**

pH	10-13.5
Density	2187 kg/m <sup>3</sup>
Particle size	2-100µm
Specific gravity	3.15
Moisture content	25.6%
Dry density	1.619 gm/cm <sup>3</sup>
Liquid limit	24.75%
Plastic limit	17.5%
Plasticity index	7.25%
IS classification	ML (silt of low plasticity)
Volumetric shrinkage	1.6%
Linear shrinkage	5.26%
Permeability	1-5 x 10 <sup>-4</sup> m/day

Table 3 Properties of red mud

**VI HEALTH IMPACTS ON HUMAN**

Red mud contains toxic metals like arsenic, chromium, cadmium and nickel. The mud also contains radioactive elements and is highly alkaline, caustic enough to burn skin and eyes. Long-term exposure to arsenic from drinking-water and food can cause cancer and skin lesions. It has also been associated with cardiovascular disease and diabetes. In utero and early childhood exposure has been linked to negative impacts on cognitive development and increased deaths in young adults.

Adverse health effects associated with Cr(VI) exposure include occupational asthma, eye irritation and damage, perforated eardrums, respiratory irritation, kidney damage, liver damage, pulmonary congestion, upper abdominal pain, nose irritation and damage, respiratory cancer and skin irritation.

Breathing high levels of cadmium damages people's lungs and can cause death. Exposure to low levels of cadmium in air, food, water, and particularly in tobacco smoke over time may build up cadmium in the kidneys and cause kidney disease and fragile bones. Cadmium is considered a cancer-causing agent.

Nickel (Ni) is a hard, silvery-white metal that may cause irritation to the skin. Exposure can harm

the lungs, stomach, and kidneys. Exposure to nickel may lead to cancer.

**VIIDISPOSAL TECHNIQUES**

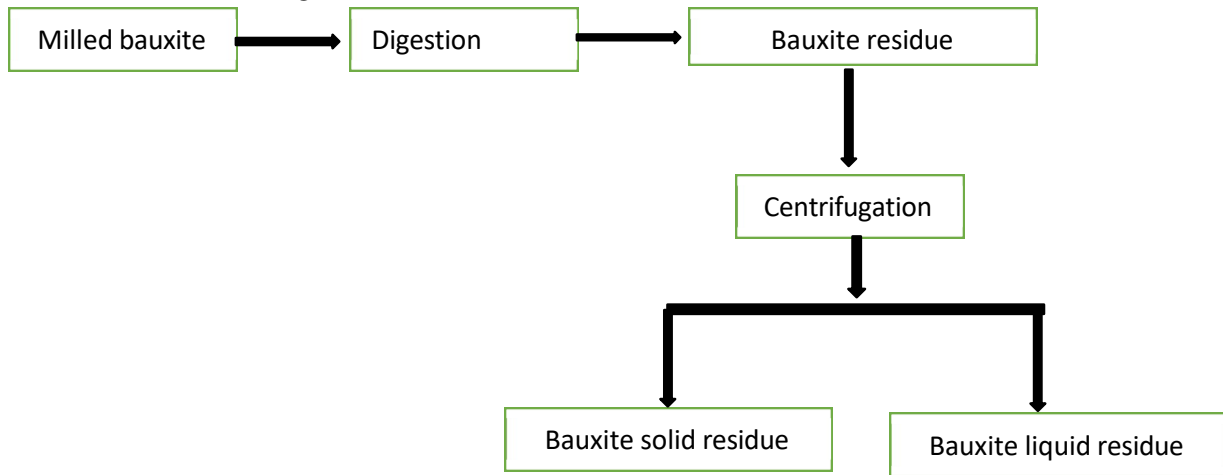


Fig 4 Separation of liquid from bauxite

The environmental risk associated with the disposal of red mud is due to high alkalinity, presence of radionuclides and toxic elements. The traditional disposal technique includes slurry disposal, dry mud disposal, sea or lake disposal, closed cycle disposal, modified closed cycle disposal methods.

Sea disposal: the slurry from the Bayer process is directly disposed to the sea through the pipeline. This may reduce the land pollution, but it increases the turbidity level in sea due to the formation of colloidal aluminium and magnesium compounds.

Conventional wet disposal: by dewatering process the sodium hydroxide is separated from the red mud slurry. Due to the seepage

**VIII UTILIZATION OF RED MUD**

Now a days red mud is effectively used in engineering field. The red mud contains sufficient amount of silica, alumina, iron oxides so it can be used in the cement of alkaline content the groundwater got contaminated.

Dry disposal: The slurry is dewatered and dried up to 48-58% of solids. This reduces the area of disposal. In India mostly dry disposal is used.

manufacturing. And in concrete red mud can be partially replaced in place of cement. In the wastewater treatment the red mud act as a adsorbent for the removal of heavy metals. It can be also used in brick manufacturing.

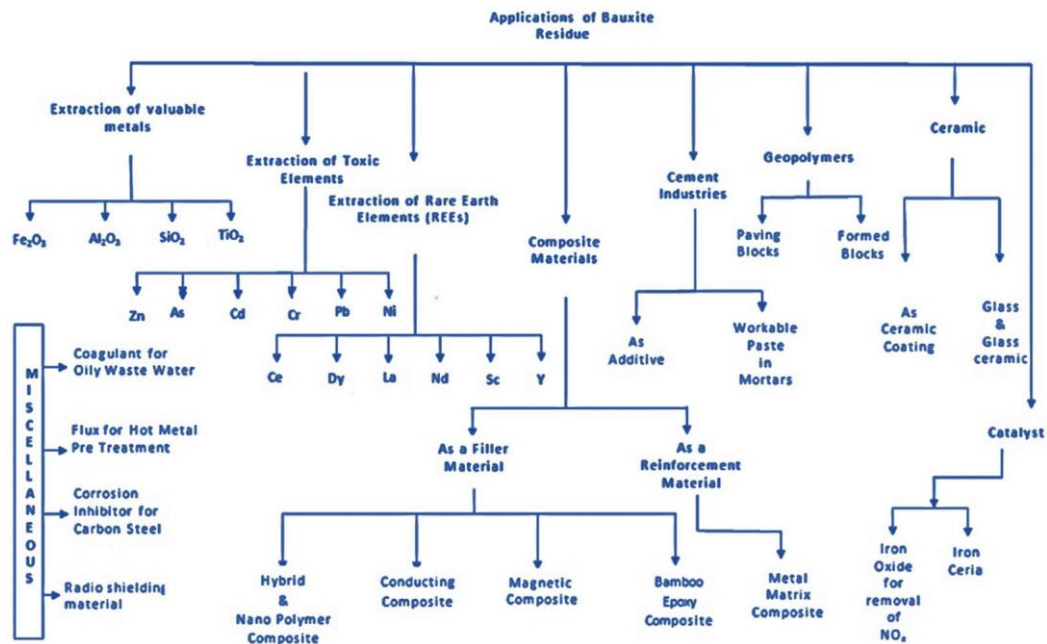


Fig 5 Utilization methods

## IX CONCLUSION

In this review article the production process and characteristics of red mud is discussed. As the production of alumina continues, the red mud accumulation will also increase. 1 ton of alumina produces 1.5 tons of red mud. In India, it is mostly disposed in dry form in open spaces, allotted for the disposal. It possesses severe negative impacts on environment and on human health. To minimize the impact, we could consider the conversion of this element into another form rather than disposing it to open spaces or oceans. It is best to utilize this waste as a raw material for another process. By using it in cement and brick manufacturing as a resource, the waste exchange process can be effectively done. Due to its absorptive property, it can be even used to absorb heavy metals in wastewater treatment process. Still, a deep and continuous research work should be carried out for the termination of the problem caused by red mud.

## REFERENCE

1. Patel, S., & Pal, B. K. (2015). Current status of an industrial waste: red mud an overview. *Ijltemas*, 4(8), 1-16.
2. Samal, S., Ray, A. K., & Bandopadhyay, A. (2013). Proposal for resources, utilization and processes of red mud in India—a review. *International Journal of Mineral Processing*, 118, 43-55.
3. Ke, W., Zhang, X., Zhu, F., Wu, H., Zhang, Y., Shi, Y., ... & Xue, S. (2021). Appropriate human intervention stimulates the development of microbial communities and soil formation at a long-term weathered bauxite residue disposal area. *Journal of Hazardous Materials*, 405, 124689.
4. Xue, S., Zhu, F., Kong, X., Wu, C., Huang, L., Huang, N., & Hartley, W. (2016). A review of the characterization and revegetation of bauxite residues (Red mud). *Environmental Science and Pollution Research*, 23, 1120-1132.
5. Borra, C. R., Blancpain, B., Pontikes, Y., Binnemans, K., & Van Gerven, T. (2016). Recovery of rare earths and other valuable metals from bauxite residue (red mud): a review. *Journal of Sustainable Metallurgy*, 2, 365-386.

6. Di Mare, M., Monteiro, V. N., Brial, V., Ouellet- Plamondon, C. M., Fortin, S., Tsesmelis, K., & Rosani, D. (2021). A calculator for valorizing bauxite residue in the cement industry. *Cleaner Materials, 1*, 100009.
7. Verma, A. S., Suri, N. M., & Kant, S. (2017).
8. Applications of bauxite residue: A mini- review. *Waste Management & Research, 35*(10), 999-1012.
9. Liu, Y., & Naidu, R. (2014). Hidden values in bauxite residue (red mud): Recovery of metals. *Waste management, 34*(12), 2662-2673.
10. Xue, S. G., Jiang, Y. F., & Zhu, F. (2022).
11. Ecological Disposal and Large-scale Utilization of Bauxite Residue: A Long way to go. *Bulletin of Environmental Contamination and Toxicology, 109*(1), 1-2.
12. Agrawal, A., Sahu, K.K., Pandey, B.D., (2004). Solid waste management in non-ferrous industries in India. *Resources, Conservation and Recycling 42* 99–120.
13. Akhurst, D. J., Jones, G. B., Clark, M., & McConchie, D. (2006). Phosphate removal from aqueous solutions using neutralised bauxite refinery residues (Bauxsol™). *Environmental chemistry, 3*(1), 65-74.
14. Altundogan, H. S., Altundogan, S., Tümen, F., & Bildik, M. (2002). Arsenic adsorption from aqueous solutions by activated red mud. *Waste management, 22*(3), 357-363.
15. Altundogan, H.S., Altundogan, S., F. Tüme F., and Bildik, M., (2000). Arsenic removal from aqueous solutions by adsorption on red mud, *Waste Manage. (20)*, pp. 761–767.
16. Amritphale, S.S., Anshul, Avneesh, Chandra, Navin, Ramkrishnan, N., (2007). A novel process for making radiopaque materials using bauxite-red mud. *J. Eur. Ceram. Soc. 27*, 1945–1951.
17. Apak R., Tütem E., Hügül M., and Hizal J., (1998). Heavy metal cation retention by unconventional sorbents (red muds and fly ashes), *Water Res. 32*, pp. 430–440.
18. Atasoy A., (2011). Reduction of Ferric Oxides In The Red Mud by The Aluminiothermic Process. 6th International Advanced Technologies Symposium (IATS'11), 16-18 May 2011, Elazığ, Turkey.
19. Bhatnagar A., Vítor J.P. Vilar, Cidália M.S. Botelho and Rui A.R. Boaventura., (2011). A review of the use of red mud as adsorbent for the removal of toxic pollutants from water and wastewater. *Environmental Technology, Vol. 32, No. 3, February 2011*, 231–249.
20. Bhattacharjee S.C., Pal D.K, and Philipose C.I., (1987). 'Bauxite Tailings - Red Mud', A.S.Wagh and P. Desai, Eds. The Jamaican Bauxite Institute and the University of West Indies, Kingston , pp. 73.
21. Snars, K., Gilkes, R.J., (2009). Evaluation of bauxite residues (red muds) of different origins for environmental applications. *Applied Clay Science ;( 46):13–20*.
22. Srikanth, S., Ray, A.K., Bandopadhyay, A., Ravikumar, B., Jha, A., (2005). Phase constitution during sintering of red mud and red mud-fly ash mixtures. *J. Am. Ceram. Soc. 88 (9)*, 2396–2401.
23. Srikanth S, Ray AK, Bandopadhyay A, Ravikumar B., (2005). Phase constitution during sintering of red mud and red mud-fly ash mixtures. *Journal of the American Ceramic Society. (88):2396–2401*.  
Sutar H., Mishra S.C., Sahoo S.K., chakraverty
24. A.P. and Maharana H.S., (2014). Progress of Red Mud Utilization: An Overview. *American Chemical Science Journal. 4*(3): 255-279.
25. Sun YF, Dong FZ, Liu JT., (2009). Technology for recovering iron from red mud by Bayer process (In Chinese). *Met. Mine ;( 9):176–178*.
26. Szirmai, E., Babusek, S., Balogh, G., Nedves, A, Horvath, G., Lebényi, Z., Pinter, J., (1991).

27. —Method for the Multistage, Waste-free Processing of Red Mud to Recover Basic Materials of Chemical Industry, US Patent 5,053,144, Oct.
28. Vaclavikova, M., Misaelides P., Gallios G., Jakabsky S., and Hredzak S., (2005). Removal of cadmium, zinc, copper and lead by red mud, an iron oxides containing hydrometallurgical waste, *Stud. Surf. Sci. Catal.* 155, pp. 517–525.
29. Reddy, P. S., Reddy, N. G., Serjun, V. Z., Mohanty, B., Das, S. K., Reddy, K. R., & Rao, B. H. (2021). Properties and assessment of applications of red mud (bauxite residue): current status and research needs. *Waste and Biomass Valorization*, 12, 1185-1217. <https://doi.org/10.1007/s12649-020-01089-z>
30. Vachon, P., Tyagi, R.D., Auclair, J.C., Wilkinson, K.J., (1994). Chemical and biological leaching of aluminum from red mud. *Environmental Science and Technology*. ;( 28):26–30.
31. Varnavas, S.P., Achilleopoulos P.P., (1995). Factors controlling the vertical and spatial transport of metal-rich particulate matter in seawater at the outfall of bauxitic red mud toxic waste, *The Science of the Total Environment*, 175 , 199-205.
32. Vaselinka V. Grudic, Dina Peric, Nada Z. Blagojevic, Vesna L. Vukasinovic- Pesic, Snezana Brasanac, Bojana Mugosa., (2013). Pb (II) and Cu (II) sorption from Aqueous solutions using Activated Red mud. Evaluation of Kinetic, Equilibrium and Thermodynamic Models. *Pol. J. Environ. Stud.* Vol. 22, No. 2 , 377-385.
33. Vidyasagar, P., (1996). Red mud separation in alumina industry for cleaner environment. In: Bhima Rao, R., Ansari, M.I. (Eds.), *Solid Liquid Separation in Mineral and Metallurgical Industries, Selected Papers contributed for the National Seminar on Solid Liquid Separation in Mineral and Metallurgical Industries*, Bhubaneswar, India, Nov. 27–28. Publisher: Indian Institute of Mineral Engineers, Bhubaneswar Chapter, Bhubaneswar, India, pp. 30–56
34. Wang, H.M., (2011). The comprehensive utilization of red mud (In Chinese). *Shanxi Energy Conserve.* ;( 11):58–61.
35. Wei N., Luan Z-K., Wang J., Shi L., Zhao Y., and Wu J.W., (2009). Preparation of modified red mud with aluminium and its adsorption characteristics on fluoride removal, *Chin. J. Inorg. Chem.* 25, pp. 849–854.
37. Yang, J.K., Chen, F., Xiao, B., (2006).
38. Engineering application of basic level materials of red mud high level pavement (In Chinese). *China Munic. Eng.* ;( 5):7–9.
39. Yang, L.G., Yao Z.L., Bao, D.S., (1996).
40. Pumped and cemented red mud slurry filling mining method (In Chinese). *Mining Res. Develop.* ;( 16):18–22.
41. Zhang S., Liu C., Luan Z., Peng X., Ren H., and Wang J., (2008). Arsenate removal from aqueous solutions using modified red mud, *J. Hazard. Mater.* 152, pp. 486–492.
42. Zhu, C., Luan Z., Wang Y., Shan X., (2007).
43. Removal of cadmium from aqueous solutions by adsorption on granular red mud (GRM). *Sep. Purif. Technol.*, 57, 161.
44. Zouboulis, A.I., Kydros, K.A., (1993). Use of red mud for toxic metals removal the case of nickel.
45. *J. Chem. Technol.* 58, 95–101.
46. 80. Nie, Q., Li, Y., Wang, G., & Bai, B. (2020).
47. Physicochemical and microstructural properties of red muds under acidic and alkaline conditions.

- Applied Sciences, 10(9), 2993.  
48. doi:10.3390/app10092993
49. Alam, S., Das, B. K., & Das, S. K. (2018).
50. Dispersion and sedimentation characteristics of red mud. *Journal of Hazardous, Toxic, and Radioactive Waste*, 22(4), 04018025. DOI: 10.1061/(ASCE)HZ.2153-5515.0000420
51. Abhilash, S. S., Meshram, P., Pandey, B. D., Behera, P. K., & Satpathy, B. K. (2014). Red Mud: a secondary resource for rare earth elements. In *International bauxite, alumina and aluminium symposium, The IBAAS Binder (Vol. 3, pp. 148- 162)*.
52. Feng, Y., & Yang, C. (2018). Analysis on physical and mechanical properties of red mud materials and stockpile stability after dilatation. *Advances in Materials Science and Engineering*, 2018. <https://doi.org/10.1155/2018/8784232>
53. Li, X., Zhang, Q., Mao, S., Li, L., & Wang, J. (2019). Study on the preparation and fracture behavior of red mud-yellow phosphorus slag-based concrete. *Advances in Materials Science and Engineering*, 2019.
54. <https://doi.org/10.1155/2019/4690802>
55. Wang, P., & Liu, D. Y. (2012). Physical and chemical properties of sintering red mud and Bayer red mud and the implications for beneficial utilization. *Materials*, 5(10), 1800-1810.
56. doi:10.3390/ma5101800
57. Joseph, C. G., Taufiq-Yap, Y. H., Krishnan, V., & Puma, G. L. (2020). Application of modified red mud in environmentally-benign applications: A review paper. *Environ. Eng. Res*, 25(6), 795-806. <https://doi.org/10.4491/eer.2019.374>
58. [44]H.-G. SCHWARZ, Aluminum Production and Energy, *Encyclopedia of Energy*, Volume 1. 2004
59. Shanmugavel, R., Jayamani, M., Nagarajan, R., Irullappasamy, S., Cardona, F., & Sultan, M. T. H. (2016, October). Processing and characterization of redmud reinforced polypropylene composites. In *IOP Conference Series: Materials Science and Engineering (Vol. 152, No. 1, p. 012053)*. IOP Publishing.
60. Paramguru, R. K., Rath, P. C., & Misra, V. N. (2004). Trends in red mud utilization—a review. *Mineral Processing & Extractive Metall. Rev.*, 26(1), 1-29.
61. Venkatesh, C., Nerella, R., & Chand, M. S. R. (2021). Role of red mud as a cementing material in concrete: A comprehensive study on durability behavior. *Innovative Infrastructure Solutions*, 6(1), 1-14. <https://doi.org/10.1007/s41062-020-00371-2>
62. Babu, U. R., & Kondraivendhan, B. (2019, October). Analysis of corrosion performance of rebar in red mud blended concrete. In *IOP Conference Series: Materials Science and Engineering (Vol. 652, No. 1, p. 012025)*. IOP Publishing
63. Bellum, R. R., Venkatesh, C., & Madduru, S. R.
64. C. (2021). Influence of red mud on performance enhancement of fly ash-based geopolymer concrete. *Innovative Infrastructure Solutions*, 6(4), 1- 9. <https://doi.org/10.1007/s41062-021-00578-x>
65. Samal, S., Ray, A. K., & Bandopadhyay, A. (2013). Proposal for resources, utilization and processes of red mud in India—a review. *International Journal of Mineral Processing*, 118, 43-55.
66. <http://dx.doi.org/10.1016/j.minpro.2012.11.001>
67. Sutar, H., Mishra, S. C., Sahoo, S. K., & Maharana, H. S. (2014). Progress of red mud utilization: An overview.



68. Verma, A. S., Suri, N. M., & Kant, S. (2017).
69. Applications of bauxite residue: A mini-review. *Waste Management & Research*, 35(10), 999-1012. <https://doi.org/10.1177/0734242X1772>
70. Revathy, T. R., Palanivelu, K., & Ramachandran, A. (2017). Sequestration of carbon dioxide by red mud through direct mineral carbonation at room temperature. *International Journal of Global Warming*, 11(1), 23-37.
71. Sundarakannan, R., Arumugaprabu, V., Manikandan, V., & Johnson, R. D. J. (2020). Tribo performance studies on redmud filled pineapple fiber composite. *Materials Today: Proceedings*, 24, 1225-1234.
72. Rai, S., Wasewar, K., Mukhopadhyay, J., Yoo,
73. C. K., & Uslu, H. (2012). Neutralization and utilization of red mud for its better waste management. *World*, 6, 5410.
74. Amrutha, M., & Ramakrishna, M. (2018). RED MUD AS CONSTRUCTION MATERIA BY USING BIOREMEDIATION. *RED*, 5(02).
75. Ribeiro, D. V., Labrincha, J. A., & Morelli, M.
76. R. (2011). Chloride diffusivity in red mud-ordinary Portland cement concrete determined by migration tests. *Materials Research*, 14, 227-234. DOI: 10.1590/S1516-14392011005000026